
BURNING QUESTIONS ABOUT BOILERS

*High Performance Boilers
&
Boiler Controls*

peaker

*Energy Engineering Services for
Massachusetts
& the Northeast Since 1990*

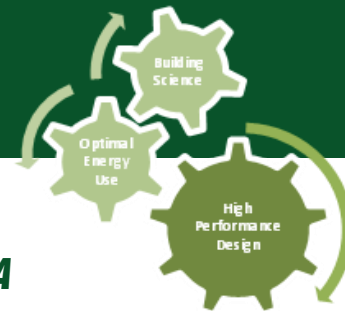


BALES ENERGY ASSOCIATES

Bart Bales PE MSME
Senior Engineer

Phone: 413-342-9352
bart.balesenergy@gmail.com

BEA



uestions to be
sked & answered



- ◆ What do you get when you cross a
Lovesick elephant with a rhinoceros?



-
- ◆ Is condensing latent heat out of the combustion gases of a boiler desirable?
 - ◆ Why are we interested in the return water temperature for a boiler system?
 - ◆ What is “excess air” and why do boilers need to operate with some “excess air”?

Measured on bus in efficiency?



Measured combustion efficiency?

- ◆ The percent of the thermal energy in the fuel burned is lost in the stack gases

here is the energy going that is not in the stack gases?

???

Where is the energy going that is not in the stack gases?

- ◆ into the boiler water
- ◆ into the mass of the metal making up the boiler
- ◆ lost from the surface of the boiler

where do you want the energy to be?

- ◆ in the boiler water or the water serving the building's distribution loop

What affects how much heat is transferred to the water?

◆ ??

What affects how much heat is transferred to the water?

- ◆ Flow rate of the water entering and passing through the boiler
- ◆ Temperature of the entering water

in routine operation of a cast iron boiler from where does this water come?

- ◆ **(assume single supply, single return boiler system)**

During routine operation of a cast iron boiler, from where does this water come?

- ◆ (assume single supply, single return boiler system)
- ◆ the return water from the building heating loop

What temperatures?

- ◆ . i ve range.

What temperatures?

- ◆ Give range.
- ◆ ... to ...
- ◆ Why is the low end of the range so high?

What temperatures?

- ◆ Wide range.
- ◆ ... to ...
- ◆ Why is the low end of the range so high?
- ◆ To avoid condensation of “icky” things from the combustion gases

Why is the low end of the range so high?

- ◆ to avoid condensation of “icky” things from the combustion gases

Why is the low end of the range so high?

◆ to avoid condensation of “icky” things from the combustion gases

◆ $\Delta T_{\text{icky}} = ?$

◆ Why is the low end of the range so high?

◆ To avoid condensation of “icky” things from the combustion gases

◆ Icky = sulfuric acid, etc.

◆ Which does what?

Why is the low end of the range so high?

- ◆ to avoid condensation of “icky” things from the combustion gases
- ◆ icky = sulfuric acid, etc.
- ◆ which does what?
- ◆ corrodes cast iron of boiler and galvanized metal of the exhaust stack.

condensing oiler

- ◆ In routine operation, what temperatures may the return water be?

condensing oiler

In routine operation, what temperatures
may the return water be?

... to... or is the high up to ...?

Why can the low temperature be lower in a condensing boiler than a standard boiler?

Why can the low temperature be lower in a condensing boiler than a standard boiler?

int: materials

Why can the low temperature be lower in a condensing boiler than a standard boiler?

hint: materials

Condensing boilers are constructed with stainless steel components which can tolerate exposure to the acidic condensate.

risk question

- ◆ In routine operation of a condensing boiler, can return water temperatures be as high as T_{sat} ?

rick question

- ◆ In routine operation of a condensing boiler, can return water temperatures be as high as ... ?
- ◆ **answer:** In a boiler designed for condensing operation, the boiler may operate with return water temperatures higher than ... but ... ?

Answer

- ◆ In a boiler designed for condensing operation, the boiler may operate with return water temperatures higher than . . .
- ◆ But, when so operated, it will not actually be condensing because?

Answer

- ◆ In a boiler designed for condensing operation, the boiler may operate with return water temperatures higher than . . .
- ◆ But, when so operated, it will not actually be condensing because: **the latent heat in the exhaust requires lower stack temperatures to condense.**

More on Condensing Boilers later

- ◆ Our second speaker, Roger Harris, will go into great detail about condensing boilers in his presentation in the second half of this session.

combustion

- ◆ What determines how much thermal energy is left in the stack gases?
?

combustion

- ◆ **What determines how much thermal energy is left in the stack gases?**

How hot the fluid in the stack is

How much total flow there is through the stack

- ◆ **What's in the gases that go up the stack?**

What determines how much thermal energy is left in the stack gases?

- ◆ How hot the fluid in the stack is
- ◆ How much total flow there is through the stack
- ◆ **What's in the gases that go up the stack?**
- ◆ Combustion products, excess air (nitrogen, oxygen), water vapor

excess air

- ◆ why is there excess air?

excess air?

- ◆ why do we need to have a little excess air?
- ◆ why have more than the stoichiometric amount?

excess air?

- ◆ why do we need to have a little excess air?
- ◆ why have more than the stoichiometric amount?
- ◆ (To avoid having to pronounce stoichiometric?)

excess air?

- ◆ Why do we need to have a little excess air?
- ◆ Because the mixing and burning of fuel and air is imperfect
- ◆ What happens to boiler performance when there is too much excess air?

Too much excess air?

- ◆ What happens to a boiler when there is too much excess air?
- ◆ Reduces stack temperature by diluting the combustion gases with excess room air,
- ◆ Reduces combustion efficiency

oo . i t t l e x c e s s i r ?

?

oo . i t t l e x c e s s i r ?

- ◆ Soot, smoke,
- ◆ Carbon monoxide, other bad things

art ...:
val uat i ng t he avi ngs
chi eved y
ur ner ont r ol s

Goals

- ◆ must know theory basics
- ◆ describe how burners work
- ◆ describe the
 - > limitations of mechanical linkages
 - > capabilities of electric actuator controls
- ◆ discuss specific burner control systems
- ◆ measuring performance & gas usage
 - > existing case

ombust i on heor y asi cs

- ◆ ombust i on?

combustion theory basics

- ◆ combustion?
 - > combining (burning) a fuel with air resulting in hot exhaust gases
- ◆ desirable heat transfer?

combustion theory aspects

- ◆ combustion?
 - > combining (burning) a fuel with air resulting in hot exhaust gases
- ◆ desirable heat transfer?
 - > heat transfer from gas to the walls of the tubes and then to the boiler water
- ◆ what then happens to the water?

combustion

- ◆ combustion?
 - > combining (burning) a fuel with air resulting in hot exhaust gases
- ◆ desirable heat transfer?
 - > heat transfer from gas to the walls of the tubes and then to the boiler water
- ◆ what then happens to the water? it gets hot and may boil...

- ◆ must increase
-
-

- ◆ what then happens to the water? it gets hot and may boil...
- ◆ what determines whether it boils or not?

combustion thermodynamics

- ◆ What then happens to the water? It gets hot and may boil...
- ◆ What determines whether it boils or not?
 - > How much heat is supplied
 - > Resulting temperature and pressure in the boiler
- ◆ Why does a hydrothermal boiler not make steam?

combustion thermodynamics

- ◆ What then happens to the water? It gets hot and may boil...
- ◆ What determines whether it boils or not?
- ◆ How much heat is supplied
- ◆ Resulting temperature and pressure in the boiler
- ◆ Why does a hydrothermal boiler not make steam? High pressure

omput i on heory asi cs

- ◆ hat i s comput i on ef f i ci ency?

combustion theory basics

- ◆ what is combustion efficiency?
 - > useful energy output / total energy input
- ◆ what is the useful energy in this case?

combustion theory basics

- ◆ what is combustion efficiency?
 - > useful energy output / total energy input
- ◆ what is the useful energy in this case?
 - > the hot water or steam produced
- ◆ where does the rest of the energy go?

combustion theory basics

- ◆ what is combustion efficiency?
 - > useful energy output / total energy input
- ◆ what is the useful energy in this case?
 - > the hot water or steam produced
- ◆ where does the rest of the energy go?
 - > elsewhere.
 - > uh?

combustion theory basics

- ◆ what is combustion efficiency?
 - > useful energy output / total energy input
- ◆ what is the useful energy in this case?
 - > the hot water or steam produced
- ◆ where does the rest of the energy go?
 - > somewhere.
 - > uh?
 - > it stays where it has been which is where?

combustion thermodynamics

- ◆ where does the rest of the energy go?
 - > nowhere.
 - > uh?
 - > it stays where it has been, in the hot gases
- ◆ what do we do with these hot gases?

combustion chemistry

- ◆ where does the rest of the energy go?
 - > nowhere.
 - > uh?
 - > it stays where it has been, in the hot gases
- ◆ what do we do with these hot gases?
 - > exhaust them up the chimney

combustion theory aspects

- ◆ Why exhaust hot gases up the chimney because?
 - > Multiple choice:
 - _ . It is cheaper to make boilers that have hot exhaust
 - _ . Fuel waste is good for the economy
 - _ . The boiler is limited by thermodynamic laws & there must always be some waste heat
 - _ . If the gases were cooler, they would make the chimney “yucky” and the boiler to “croak”

Why must you hear y asi cs

- ◆ Why exhaust hot gases up the chimney because?
 - > Multiple choice:
 - _ . It is cheaper to make boilers that have hot exhaust
 - _ . Fuel waste is good for the economy
 - _ . The boiler is limited by thermodynamic laws & there must always be some waste heat
 - _ . If the gases were cooler, they would make the chimney “yuccify” and the boiler to “croak”
 - _ . Answer: , , and a little .

must monitor chemistry

- ◆ if the gases were cooled enough
 - > the acidic parts of the exhaust will condense
 - _ on the chimney
 - _ on the boiler heat transfer surfaces
 - > and damage them

Combustion theory aspects

- ◆ To ensure complete combustion of the fuel used, combustion chambers are supplied with excess air.
- ◆ Excess air increases the amount of oxygen and the probability of combustion of all fuel.
- ◆ When fuel and oxygen in the air are in perfectly balance - the combustion is said to be stoichiometric.
- ◆ The combustion efficiency will increase with increased excess air, until the heat loss in the excess air is larger than the heat provided by more efficient combustion.
- ◆ Typical excess air to achieve highest efficiency for different fuels are
 - ◆ . - % for natural gas
 - ◆ . - % for fuel oil

Robustness in theory and practice

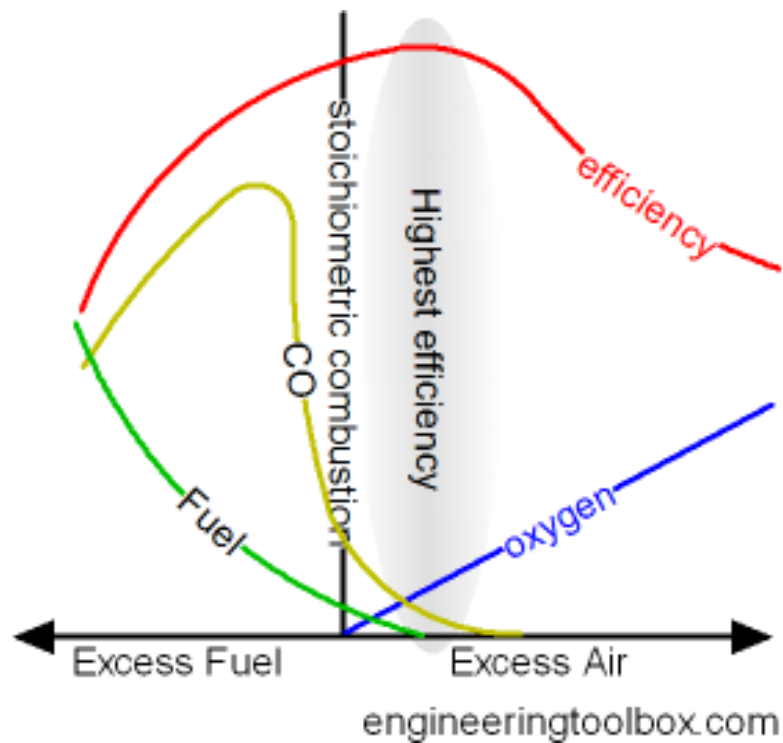
- ◆ approximate minimum as empirates to avoid orrosion problems

> atural as-

> il uel – ypi cal sul fur $\% > . \%$

> il uel – ow sul fur $\% < . \%$

combustion theory aspects



combustion theory basics

- ◆ carbon dioxide - CO_2 - is a product of the combustion and the content in the flue gas is an important indication of the combustion efficiency.

after combustion:

optimal content of carbon dioxide - CO_2 -

~% for natural gas

- ~% for light oils.

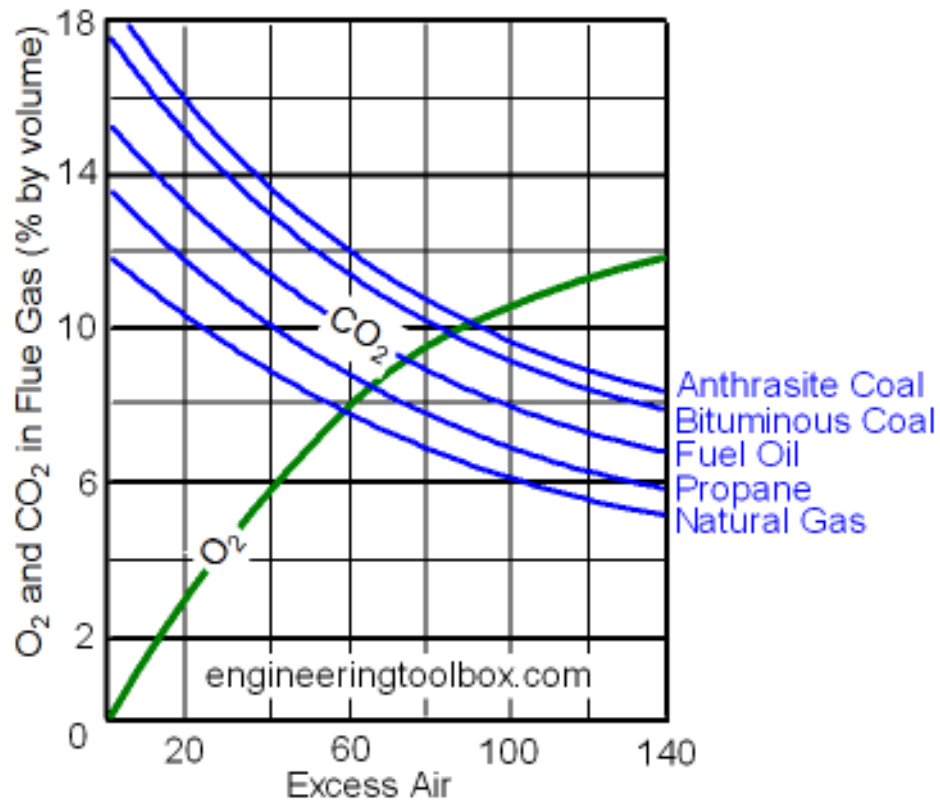
combustion theory aspects

- ◆ what is the relationship of the stack gas percentages of:
- ◆ ?
- ◆ oxygen?

Must know theory basics

- ◆ What is the relationship of the stack gas percentages of :
 - > ?
 - > oxygen?

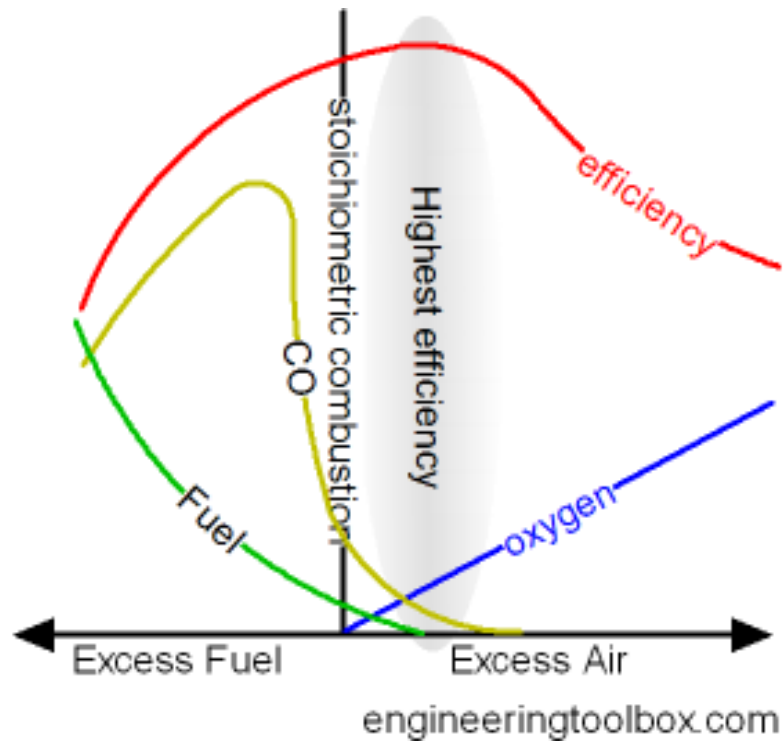
combustion theory basics



ombustion theory basics

- ◆ take a look at the combustion chart again
 - > with regard to efficiency:
 - _ what is the impact of oxygen levels?
 - _ at what level of excess oxygen become maximum?
 - _ what happens to efficiency when there is too little oxygen?

combustion theory basics



combustion efficiency

Normal combustion efficiencies for natural gas at different amounts of excess air and flue gas temperatures are indicated below:

Combustion Efficiency (%)						
Excess %		Flue Gas Temperature (°F)				
Air	Oxygen	200	300	400	500	600
9.5	2.0	85.4	83.1	80.8	78.4	76.0
15	3.0	85.2	82.8	80.4	77.9	75.4
28.1	5.0	84.7	82.1	79.5	76.7	74.0
44.9	7.0	84.1	81.2	78.2	75.2	72.1
81.6	10.0	82.8	79.3	75.6	71.9	68.2

Turner controls for controlling air & fuel

- ◆ Functions?
 - > optimize air/fuel ratio
 - how?

controller roles for controlling air & fuel

- ◆ functions?
 - > optimize air/fuel ratio
 - _ ensure complete fuel combustion
 - _ maintain optimal excess-air levels
 - _ maintain target boiler temperature or pressure
 - leadjust respond quickly to changes in external conditions
 - _ barometric pressure
 - _ wind
 - _ weather

turner controls for controlling i r & fuel

- ◆ Id way:
 - > adjustable linkages
 - _ connecting rods regulate
 - fuel valve
 - i r damper opening
 - _ unstable efficiencies:
 - cannot be optimized for all conditions
 - _ imprecise
 - do not track the same when
 - _ ramping up and ramping down

Turner Controls for controlling air & fuel

- ◆ New way:
 - > linkage-less actuators with microprocessors
 - Electric actuators provide independent control of
 - fuel valve
 - air damper opening
 - precise
 - rack the same when
 - ramping up and ramping down

turner controls for controlling ir & fuel

- ◆ add stack monitoring capability
 - > measure $\dot{m}_f, \dot{m}_a, X, X$
 - > microprocessor continually optimizes for all conditions
 - > table efficiencies across the firing range

Superduper Controls for Controlling IR & Fuel

- ◆ Question:
 - > How do you have this superduper control? How do you apply it?
 - > How do we set it up to maintain minimum stack temperatures no matter what?
 - > How well can we do?
 - Assume a typical cast-iron boiler
 - What efficiency can we target?

Burner controls for controlling \dot{m}_r & fuel

- ◆ Question: with these burner controls,
> what efficiency can we target?

— %
— %
— %
— %
— %
— %
— %
— %
— % or less

Burner controls for controlling \dot{r} & fuel

- ◆ Question: with these burner controls,
 - > what stack temperature can we target?



Burner controls for controlling air & fuel

- ◆ Question: with these burner controls,
> what % level can we target?

- % or less

- .%

- %

- .%

- %

- .%

- .%

- .%

- % or more

Burner controls for controlling air & fuel

- ◆ Question: with these burner controls,
 - > What % savings can we achieve?
 - % or less
 - %
 - %
 - %
 - %
 - %
 - %
 - %
 - 8% or more

Turner controls for controlling air & fuel

The burning question:

With these controls,

- ◆ What % savings can we achieve?

- > What do you need to know to estimate the savings in terms?

Turner Controls for controlling air & fuel

The burning question: with these
controls:

- ◆ What %savings can we achieve?
- ◆ What do you need to know to estimate the savings?
 - > How many therms is the boiler using now?

burner controls for controlling air & fuel

the burning question:

with these controls,

- ◆ what %savings can we achieve?
- ◆ what do you need to know to estimate the savings?
 - > how many therms is the boiler using now?
 - > what type of boiler is it?
 - > what type of burner controls does it have now?
- ◆ what is the key single-most important question to answer about the existing boiler?

Turner Controls for controlling air & fuel

The burning question with these
controls,

- ◆ What % savings can we achieve?
- ◆ What is the key single-most important question to answer about the existing boiler?

Answer: What efficiency is it operating at now?

You can't evaluate savings meaningfully unless you know the answer to this question.

Current controls for controlling air & fuel

The new (grammatically optimized) burning question:

What efficiency is it operating **now?**

How do we find out?

Current controls for controlling air & fuel

The new (grammatically optimized) burning question:

What efficiency is it operating **now?**

How do we find out?
Test it.

Questions?

???

Thank you.

*Energy Engineering Services for
Massachusetts
& the Northeast Since 1990*



BALES ENERGY ASSOCIATES

Bart Bales PE MSME
Senior Engineer

Phone: 413-342-9352
bart.balesenergy@gmail.com

BEA

